# QM with Atomic Physics 

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## Formalism

| symbols | meaning |
| :---: | :---: |
| $\Psi(\mathrm{x}, \mathrm{t})$ | Probability function of position and time, defines the probability of a particle occupying a position at a given time |
| $\Psi *$ | The complex conjugate (replace $i$ with -i) |
| $\begin{gathered} \int_{-\infty}^{\infty} \Psi^{*}(x, t) \Psi(x, t) d x=1 \\ \int_{-\infty}^{\infty}\|\Psi(x, t)\|^{2} d x=1 \end{gathered}$ | Normalization: the probability of finding a particle that exists somewhere, across all possible somewheres, is 1 . <br> $\Psi$ may need to be normalized. <br> $\Psi$ goes to 0 at positive \& negative infinity. |
| $\hat{A}$ | An operator: a "rule" that acts on a function, in QM it acts on $\Psi$ ex. $\frac{d}{d x}, 2 x$ |
| $\langle A\rangle$ $\int_{-\infty}^{\infty} \Psi^{*}(x, t) \hat{A} \Psi(x, t) d x=\langle A\rangle$ | Expected value of $A$ |

# If a function $f(x)$ operated on by operator returns $c^{*} f(x)$, then $f(x)$ is an eigenfunction of that operator, with eigenvalue $c$. 

| Math | English |
| :---: | :--- |
| $\int_{-\infty}^{\infty} f(x)^{*}(\hat{A} g(x)) d x=\int_{-\infty}^{\infty}(\hat{A} f(x))^{*} g(x) d x$ | Definition of a Hermitian operator |
| $\int_{-\infty}^{\infty} f(x)^{*} g(x) d x=0$ | Eigenfunctions corresponding to different <br> eigenvalues are orthogonal. |
| $\hat{x}=x$ | Position and momentum operators <br> All other physical observable operators can be <br> derived from these, except for spin. |
| $c_{n}=-i \hbar \frac{\partial}{\partial x}$ | Cn is the probability coefficient of the function <br> being in a specific state |
| $\langle A\rangle=\sum_{-\infty}^{\infty} f_{n}(x)^{*} \Psi(x, t) d x$ | The expectation value of A can be calculated if <br> the eigenvalues and coefficients are known |

## Dirac delta notation

- aka bra(c)ket notation
- $\mid B>$ is " $B$ ket"
- $\langle B|$ is "bra A"
- $\langle a \mid b\rangle:=\langle b \mid a\rangle^{*}$
- "A dagger", which is the complex conjugate of the operator A
- If $\langle a \mid \hat{A} b\rangle:=\left\langle\hat{A}^{\dagger} b \mid a\right\rangle^{*}$ is true, then A is Hermitian
- $\langle f \mid g\rangle:=\int_{-\infty}^{\infty} f(x)^{*} g(x) d x$


## Schrodinger equation

## Equation

## Explanation

$i \hbar \frac{\partial}{\partial t} \Psi(x, t)=\widehat{H} \Psi(x, t)=E_{n} \Psi(x, t)$
$\widehat{H}=\frac{\hat{p}^{2}}{2 m}+\hat{V}(x)=-\frac{\hbar^{2}}{2 m} \frac{\partial^{2}}{\partial x^{2}}+\hat{V}(x)$
$\Psi(x, t)=e^{-\frac{i E_{n} t}{h}} \psi_{n}(x)$
Schrodinger equation, all solutions are energy states
The Hamiltonian operator

General solution to the Schrodinger equation

## What to know about the wavefunction

- $\psi_{n}$ for different eigenvalues are orthogonal (their integral equals zero)
- $\psi$ is continuous, and the derivative usually also so
- $\psi_{0}$ is the ground state and has no nodes (place where the wavefunction vanishes)
- Even functions are symmetric about a vertical axis - varies by $\psi_{n}$


## Operators

- ... do not commute
- They are computed left to right, and can be used together $[A, B]=A B-B A$
- $[\hat{x}, \hat{p}]=i \hbar$ is the most important one
- Operators are involved in uncertainty, as in the Heisenberg uncertainty principle $\sigma_{x} \sigma_{p} \geq \frac{\hbar}{2}$, approximated by $\Delta x \Delta p \approx \hbar$ and $\Delta E \Delta t \approx \hbar$
- Also, $\sigma_{A}{ }^{2}=\left\langle A^{2}\right\rangle-\langle A\rangle^{2}$


## Quantum Harmonic Oscillator 1D

$$
\begin{gathered}
H=\frac{\hat{p}^{2}}{2 m}+\frac{1}{2} m \omega^{2} \hat{x}^{2} \\
{\left[a, a^{\dagger}\right]=1 \quad H|0\rangle=\frac{\hbar \omega}{2}|0\rangle} \\
H|n\rangle=\hbar \omega\left(n+\frac{1}{2}\right)|n\rangle, \quad n=0,1,2, \ldots \\
\left.a^{\dagger}|n\rangle=\sqrt{2}\right) \\
n+1|n+1\rangle ; \quad a|n\rangle=\sqrt{n}|n-1\rangle,
\end{gathered}
$$

$$
\langle T\rangle=\langle V\rangle=\frac{E_{n}}{2} .
$$



Horizontal axis is position, and the vertical axis is the real part (blue) or imaginary part (red) of the wavefunction. C, D, E, F, but not $\mathrm{G}, \mathrm{H}$, are energy eigenstates. H is acoherent state-a quantum state that approximates the classical trajectory.

## Quantum Harmonic Oscillator 3D

$$
\begin{gathered}
\psi_{N}(x, y, z)=\psi_{n_{1}}(x) \psi_{n_{2}}(y) \psi_{n_{3}}(z) ; \quad E_{N}=\left(N+\frac{3}{2}\right) \hbar \omega \text { with } N=n_{1}+n_{2}+n_{3} \\
\left(n_{1}, n_{2}, n_{3}\right)=(1,0,0),(0,1,0),(0,0,1)
\end{gathered}
$$

## Infinite Square Well

$$
H=-\frac{\hbar^{2}}{2 m} \frac{d^{2}}{d x^{2}}+V(x), \quad V(x)= \begin{cases}0, & 0 \leq x \leq a \\ \infty, & \text { otherwise }\end{cases}
$$

$$
\psi_{n}=\sqrt{\frac{2}{a}} \sin \left(\frac{n \pi x}{a}\right), \quad E_{n}=\frac{n^{2} \pi^{2} \hbar^{2}}{2 m a^{2}}
$$



$$
V(x, y, z)=\left\{\begin{array}{lr}
0, & 0 \leq x, y, z \leq a \\
\infty, & \text { otherwise }
\end{array}\right.
$$

## Free Particle

$$
H=-\frac{\hbar^{2}}{2 m} \frac{d^{2}}{d x^{2}} .
$$

## Delta Function

$$
H=-\frac{\hbar^{2}}{2 m} \frac{d^{2}}{d x^{2}}-A \delta(x)
$$

$$
\int_{-\epsilon}^{\epsilon}\left(-\frac{\hbar^{2}}{2 m} \frac{d^{2}}{d x^{2}} \psi(x)\right) d x-A \int_{-\epsilon}^{\epsilon} \delta(x) \psi(x) d x=E \int_{-\epsilon}^{\epsilon} \psi(x) d x
$$



$$
\int_{-\epsilon}^{\epsilon}\left(-\frac{\hbar^{2}}{2 m} \frac{d^{2}}{d x^{2}} \psi(x)\right) d x-A \int_{-\epsilon}^{\epsilon} \delta(x) \psi(x) d x=-\left.\frac{\hbar^{2}}{2 m}\left(\frac{d \psi}{d x}\right)\right|_{-\epsilon} ^{\epsilon}-A \psi(0),
$$

$$
\psi(x)=\frac{\sqrt{m A}}{\hbar} e^{-m A|x| / \hbar^{2}}, \quad E=-\frac{m A^{2}}{2 \hbar^{2}} .
$$

## Finite Square Well

$$
H=-\frac{\hbar^{2}}{2 m} \frac{d^{2}}{d x^{2}}+V(x), \quad V(x)=\left\{\begin{array}{lr}
-V_{0}, & -a \leq x \leq a \\
0, & \text { otherwise }
\end{array}\right.
$$



Wave function inside the well turning to decaying exponentials outside the well.

## Scattering States: Reflection/Transmission

$\psi(x)=\left\{\begin{array}{lr}A e^{i k x}+B e^{-i k x}, & x \leq-a \\ \text { something, } & -a \leq x \leq a \\ C e^{i k x}, & x \geq a\end{array}\right.$

$$
R=\frac{|B|^{2}}{|A|^{2}}, \quad T=\frac{|C|^{2}}{|A|^{2}} .
$$

$$
V(x)= \begin{cases}0, & x \leq 0 \\ V_{0}, & x \geq 0\end{cases}
$$



## Three Dimensions

- Momentum Operator:

$$
\begin{gathered}
\hat{p}=-i \hbar \nabla \\
\nabla=\frac{\partial f}{\partial x} \hat{x}+\frac{\partial f}{\partial y} \hat{y}+\frac{\partial f}{\partial z} \hat{z}
\end{gathered}
$$

## Hamiltonian in 3D

$$
\begin{gathered}
H=\frac{-\hbar^{2}}{2 m} \nabla^{2}+V(r) \\
\nabla^{2}=\frac{\partial^{2}}{\partial x^{2}}+\frac{\partial^{2}}{\partial y^{2}}+\frac{\partial^{2}}{\partial z^{2}}
\end{gathered}
$$

# Different Coordinates Commute with Each Other 

$$
\begin{aligned}
& {\left[\widehat{x}, \widehat{x_{x}}\right]=\left[\widehat{y}, \widehat{p_{y}}\right]=\left[\hat{z}, \widehat{p_{z}}\right]=i \hbar} \\
& {[\hat{x}, \hat{y}]=\left[\widehat{x}, \widehat{p_{y}}\right]=\left[\hat{x}, \widehat{p_{z}}\right]=0}
\end{aligned}
$$

## Spherical Wave Function

- Spherical Wave function: Product of angular and radial part

$$
\psi(r)=R(r) Y(\theta, \phi)
$$

- Separate into radial and angular part and normalize each piece separately

$$
\begin{gathered}
\int_{0}^{2 \pi} \int_{0}^{\pi}|\mathrm{Y}(\theta, \phi)|^{2} \sin \theta d \theta d \phi=1 \\
\int_{0}^{\infty}|R(r)|^{2} r^{2} d r=1
\end{gathered}
$$

## Angular Momentum

$$
\begin{gathered}
\hat{L}=\hat{r} \times \hat{p} \\
\widehat{L_{x}}=\hat{y} \widehat{p_{z}}-\hat{z} \widehat{p_{y}} \\
\widehat{L_{y}}=\hat{z} \widehat{p_{x}}-\widehat{x} \widehat{p_{z}} \\
\widehat{L_{z}}=\hat{x} \widehat{p_{y}}-\hat{y} \widehat{p_{x}}
\end{gathered}
$$

## Something to Remember For GRE

$$
\begin{aligned}
& {\left[\widehat{L_{x}}, \widehat{L_{y}}\right]=i \hbar \widehat{L_{z}}} \\
& \widehat{L^{2}}=\widehat{L_{x}}+\widehat{L_{y}}+\widehat{L_{z}}
\end{aligned}
$$

## Spherical Coordinates

$$
\widehat{L}^{2}=-\hbar^{2} \nabla_{\theta, \phi}^{2}
$$

$$
\widehat{L_{z}}=i \hbar^{2} \frac{\partial}{\partial \phi}
$$

## Eigenfunction and Eigenvalues

$$
\begin{aligned}
& Y_{l}^{m}(\theta, \phi) \\
& \widehat{L_{z}} Y_{l}^{m}=m \hbar Y_{l}^{m} \\
& \widehat{L^{2}} Y_{l}^{m}=l(l+1) \hbar^{2} Y_{l}^{m} \\
& \quad(m, l) \in \mathbb{Z}, l \geq 0 \\
& m=l, l-1, l-2, \ldots \ldots,-l
\end{aligned}
$$

## Some Facts

- Orthonormal:

$$
\begin{gathered}
\int_{0}^{2 \pi} \int_{0}^{\pi} Y_{l}^{m} * Y_{l}^{m^{\prime}} \sin \theta d \theta d \phi=\delta_{m m^{\prime}} \\
\delta_{m m^{\prime}}=\left\{\begin{array}{l}
0 \text { if } m \neq m^{\prime} \\
1 \text { if } m=m^{\prime}
\end{array}\right.
\end{gathered}
$$

- The $\phi$ dependence is always in the form $e^{i m \phi}$
- Dependence of $\theta$ is complicated and will be provided


## Hydrogen Atom

- Hamiltonian:

$$
H=\frac{-\hbar^{2}}{2 \mu} \nabla^{2}-\frac{e^{2}}{4 \pi \varepsilon_{0} r}
$$

- Bohr Radius

$$
a=\frac{4 \pi \varepsilon_{0} \hbar^{2}}{\mu e^{2}}
$$

## Energy States

- Ground state energy (Hydrogen):

$$
-E_{1}=\frac{h^{2}}{2 \mu a^{2}}=13.6 \mathrm{eV}
$$

- n state energy:

$$
-E_{n}=\frac{\pi^{2}}{2 \mu a^{2}} * \frac{1}{n^{2}} \quad n=1,2,3, \ldots .
$$

## Rydberg's Formula

$$
f \propto \frac{1}{n_{f}^{2}}-\frac{1}{n_{i}^{2}}
$$

## Useful for GRE

$$
\alpha=\frac{e^{2}}{4 \pi \epsilon_{0} \hbar c} \approx 1 / 137
$$

## Spin

Spin is a characteristic of a particle, and doesn't change!

Spin operators obey same commutation relations as angular momentum operators.

Spin can either be an integer or half-integer.
For GRE, only really need to know spin- $1 / 2$
Adding spins is not trivial:
For system of two particles (spin s and spin s')

$$
s_{t o t}=s+s^{\prime}, s+s^{\prime}-1, s+s^{\prime}-2, \ldots,\left|s-s^{\prime}\right|
$$

## Spin-1/2

$$
\begin{gathered}
S_{x}=\frac{\hbar}{2} \sigma_{x}, \quad S_{y}=\frac{\hbar}{2} \sigma_{y}, \quad S_{z}=\frac{\hbar}{2} \sigma_{z} \\
\sigma_{x}=\left(\begin{array}{ll}
0 & 1 \\
1 & 0
\end{array}\right) \quad \sigma_{y}=\left(\begin{array}{cc}
0 & -i \\
i & 0
\end{array}\right) \quad \sigma_{z}=\left(\begin{array}{cc}
1 & 0 \\
0 & -1
\end{array}\right) \\
|\Uparrow\rangle=\binom{1}{0},|\Downarrow\rangle=\binom{0}{1} \quad|\pi\rangle_{x}=\frac{1}{\sqrt{2}}\binom{1}{1},|\Downarrow\rangle_{x}=\frac{1}{\sqrt{2}}\binom{1}{-1} \\
\binom{a}{b}=a|\pi\rangle+b|\Downarrow\rangle \quad|\pi\rangle_{y}=\frac{1}{\sqrt{2}}\binom{1}{i},|\Downarrow\rangle_{y}=\frac{1}{\sqrt{2}}\binom{1}{-i}
\end{gathered}
$$

## Spin and the wavefunction

- Can think of total wavefunction as a product of a spatial wavefunction and a spin wavefunction
- Spin operators always commute with spatial operators
- If particle has spin-0, it only has a spatial wavefuntion
Note: the Hamiltonian can act on both the spin and spatial wavefunctions!


## Bosons and fermions

| Bosons | Fermions |
| :--- | :--- |
| Integer spin | Half-integer spin |
| Symmetric wavefunctions | Antisymmetric wavefunctions |
| Photons, alpha particles, deuterons | Electrons, muons, protons, neutrons |

Pauli exclusion principle:

- no 2 identical fermions can occupy the same quantum state For identical particles...
- symmetric wavefunctions: $\Phi\left(x_{1}, x_{2}\right)=\Phi\left(x_{2}, x_{1}\right)$
- antisymmetric wavefunctions: $\Phi\left(x_{1}, x_{2}\right)=-\Phi\left(x_{2}, x_{1}\right)$


## Approximation Methods in QM

Variation principle

- Approx. ground state E of system when Hamiltonian known
- For a normalized wavefunction: $\langle\psi| \widehat{H}|\psi\rangle \geq E_{\text {gs }}$
- Use trial wavefunction with adjustabie parameter, calculate and minimize with respect to parameter
Adiabatic theorem
- For particle in nth eigenstate of H , if H is slowly changed to $\mathrm{H}^{\prime}$, then particle will end up in the corresponding eigenstate
Note: will not have to apply variation principle on GRE!!


## Time-independent perturbation theory

$$
H=H_{0}+\lambda H^{\prime}
$$

- $\lambda \ll 1$, know energies and eigenfunctions of HO , so can compute corrections
- First-order:

$$
E_{n}=E_{n}^{0}+\lambda\left\langle\psi_{n}^{0}\right| H^{\prime}\left|\psi_{n}^{0}\right\rangle
$$

- Second-order:

$$
E_{n}=E_{n}^{0}+\lambda^{2} \sum_{m \neq n} \frac{\left.\left|\left\langle\psi_{m}^{0}\right| H^{\prime}\right| \psi_{n}^{0}\right\rangle\left.\right|^{2}}{E_{n}^{0}-E_{m}^{0}}
$$



- Diagonalize the perturbation in the subspace of degenerate states!

$$
H_{d e g e n}^{\prime}=\left(\begin{array}{cc}
H_{a a}^{\prime} & H_{a b}^{\prime} \\
H_{b a}^{\prime} & H_{b b}^{\prime}
\end{array}\right) \equiv\left(\begin{array}{cc}
\left\langle\psi_{a}\right| H^{\prime}\left|\psi_{a}\right\rangle & \left\langle\psi_{a}\right| H^{\prime}\left|\psi_{b}\right\rangle \\
\left\langle\psi_{b}\right| H^{\prime}\left|\psi_{a}\right\rangle & \left\langle\psi_{b}\right| H^{\prime}\left|\psi_{b}\right\rangle
\end{array}\right)
$$

## Atomic Review

## Bohr Model

- Electrons move in classical circular orbits, called energy shells or energy levels.
- Angular Momentum - L=n*ћ where n=1,2,3...
- Electrons do not radiate as they move around the nucleus.
- Classically the electron would radiate and spiral into the proton.

Angular Momentum


## Perturbations to the Hydrogen Atom

## Fine Structure

- Replacing the electron KE term in the Hamiltonian with the correct relativistic form.
- Spin-Orbit coupling between electron's orbital angular momentum and its spin.

Lamb Shift

- Splits the 2 s and 2 level with $\mathrm{j}=1 / 2$.

Spin-Spin Coupling

- Ground state of hydrogen splits.
- Depends on spins of e and p.

Useful Calculation Tip


## Shell Model and Electronic Notation

Orbitals

| Number | Symbol | Possible Values |
| :--- | :---: | :---: |
| Principal Quantum Number | $n$ | $1,2,3,4, \ldots$ |
| Angular Momentum Quantum Number | $\ell$ | $0,1,2,3, \ldots,(n-1)$ |
| Magnetic Quantum Number | $m_{1}$ | $-\ell, \ldots,-1,0,1, \ldots, \ell$ |
| Spin Quantum Number | $m_{\mathrm{s}}$ | $+1 / 2,-1 / 2$ |

$L=n \hbar \quad L^{2}=l(l+1) \hbar^{2} \quad|L|=\hbar \sqrt{ } 2$
$1 s_{2}^{2} 2 s_{4}^{2} 2 p_{10}^{6} 3 s_{12}^{2} 3 p_{18}^{6} 4 s_{20}^{2} 3 d_{30}^{10} 4 p_{36}^{6} 5 s_{38}^{2} 4 d_{48}^{10} 5 p_{54}^{6} 6 s_{56}^{2} 4 f_{70}^{14} 5 d_{80}^{10} 6 p_{86}^{6} 7 s_{88}^{2} 5 f_{102}^{14} 6 d_{112}^{10} 7 p_{118}^{6}$
Shells fill in order, preferring smaller values of I until Argon.
Noble gases are chemically inert because they have totally filled electron shells.

- Alkali metals have one "extra" electron
- Halogens have one fewer electron


## Stark Effect



Splitting of degenerate energy levels caused by E-field.
Change in Hamiltonian in a uniform electron field is given by,

- $\Delta H=e E \square r$

There is no change in the ground state energy
First states to show a first-order shift are $\mathrm{n}=2$ states.

- States with $\mathrm{m}= \pm 1$ are unaffected, but 2 s and $2 p$ states with $\mathrm{m}=0$ are split.

Energy splitting must be given of the form,

- $\Delta E=k e|E| d$, where $d$ is some length (usually $d=a_{0}$ ), and $k$ is an undetermined constant.


## Zeeman Effect

- Splitting of degenerate energy levels caused by B-field.
- Change in Hamiltonian in a magnetic field is given by,
- $\Delta H=(e / 2 m)^{*}(L+2 S) \cdot B$
- e/2m is the electron's classical gyromagnetic ratio.
- 2 in front of spin operator is because the quantum gyromagnetic ratio is twice the classical value.



## Selection Rules

- No transitions occur unless,
- $\Delta \mathrm{m}= \pm 1$ or 0
- Conservation of the z-component of angular momentum
- Photon Spin $=1 \rightarrow /=-\hbar, 0, \hbar$
- $\Delta /= \pm 1$
- Decay Pattern Example:
- $3 \mathrm{~s} \rightarrow 2 \mathrm{p} \rightarrow 1 \mathrm{~s}$


## Blackbody Radiation

$$
I(\omega) \propto \frac{h \omega^{2}}{c^{2}} \frac{1}{e^{\frac{\hbar}{k_{b} T}}-1}
$$

